Name: Key **Date:** _____

Balancing Equations

Balance the following chemical equations.

1. 1 $SiCl_4(l)$ + 2 $H_2O(l)$ → 1 $SiO_2(s)$ + 4 $HCl(aq)$

2. 2 As + 6 $NaOH$ → 2 Na_3AsO_3 + 3 H_2

3. 1 Au_2S_3 + 3 H_2 → 2 Au + 3 H_2S

4. 1 V_2O_5 + 6 HCl → 2 $VOCl_3$ + 3 H_2O

5. 3 $Hg(OH)_2$ + 2 H_3PO_4 → 1 $Hg_3(PO_4)_2$ + 6 H_2O

6. 1 SiO_2 + 4 HF → 1 SiF_4 + 2 H_2O

7. 1 Zn + 2 HCl → 1 $ZnCl_2$ + 1 H_2

8. 12 $HClO_4$ + 1 P_4O_{10} → 4 H_3PO_4 + 6 Cl_2O_7

9. 2 $N_2(g)$ + 5 $O_2(g)$ + 2 H_2O → 4 $HNO_3(aq)$

10. 2 NH_4NO_3 → 2 N_2 + 1 O_2 + 4 H_2O